# CHEM40111/CHEM40121 Molecular magnetism 8 Quantum information processing



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## Course Overview

<ol> <li>Fundamentals</li> <li>Motivation</li> <li>Origins of magnetism</li> <li>Bulk magnetism</li> </ol>	<ul> <li>5 Single-molecule magnets I</li> <li>Single-molecule magnets</li> <li>Electrostatic model</li> </ul>
<ul> <li>2 Quantum mechanics of magnetism</li> <li>Zeeman effect</li> <li>Statistical mechanics</li> <li>Magnetisation</li> <li>Magnetic susceptibility</li> </ul>	<ul> <li>6 Single-molecule magnets II</li> <li>Measuring magnetic relaxation</li> <li>Relaxation mechanisms</li> <li>Latest research</li> </ul>
<ul> <li>3 Magnetic coupling</li> <li>• Exchange Hamiltonian</li> <li>• Experimental measurements</li> <li>• Vector coupling</li> </ul>	<ul> <li>7 Magnetic resonance imaging</li> <li>Paramagnetic NMR</li> <li>Magnetic resonance imaging</li> <li>Latest research</li> </ul>
<ul> <li>4 Magnetic anisotropy</li> <li>2ero-field splitting</li> <li>Impact on properties</li> <li>Lanthanides</li> <li>Spin-orbit coupling</li> </ul>	<ul> <li>8 Quantum information processing</li> <li>Quantum information</li> <li>DiVincenzo criteria</li> <li>Latest research</li> <li>Question time</li> </ul>

### Intended learning outcomes

- 1. Explain the origin of magnetism arising from electrons in atoms and molecules using formal quantum-mechanical terms
- 2. Compare and contrast the electronic structure of metal ions in molecules and their magnetic properties, for metals across the periodic table
- 3. Select and apply appropriate models and methods to calculate molecular magnetic properties such as magnetisation, magnetic susceptibility and paramagnetic NMR shift
- 4. Deconstruct topical examples of molecular magnetism including single-molecule magnetism, molecular quantum information processing and MRI contrast agents

### Quantum information

• UP and DOWN states are classical bits

• Quantum bits (qubits) can be anywhere on the Bloch sphere

 $|\psi\rangle = \alpha |1\rangle + \beta |0\rangle$ 

• Allows quantum computers to do things classical computers cannot



### Quantum information

We talk about *coherence time* of qubits 0) - This is how long we can manipulate the state  $|\psi\rangle$  before the qubit gets messed up  $|\Psi\rangle$ θ We want this to be as long as possible! y Х 1)

### Quantum information



## Building a quantum computer

- The DiVincenzo criteria:
- 1. Scalable system with well-characterized qubits
- 2. The ability to initialize the qubits
- 3. Long coherence times
- 4. A universal set of quantum gates
- 5. The ability to readout the qubits
- Many ways to implement qubits:
  - Trapped ions
  - Superconducting circuits (D-Wave, Google, IBM)
  - Defects in diamond
  - Molecular magnets

[1] D. P. DiVincenzo, Fortschritte der Physik, 2000, 48, 771.

### NV centres

- Nitrogen-vacancy (NV) centres are defects in diamond
  - A nitrogen impurity is next to a vacancy, and is charged -1:



- N atom has three σ-bonds and a lone pair
- Vacancy has three
   "dangling" σ bonds, each
   with one electron
- An extra electron resides at the vacancy

• Overall, two electrons pair, and the other two give S = 1

#### NV centres

• The *S* = 1 state has axial ZFS, and there are also excited *S* = 1 and *S* = 0 states:



[1] M. S. Fataftah and D. E. Freedman, Chem. Commun., 2018, 54, 13773.

- A single [Tb(phthalocyanine)<sub>2</sub>] SMM is placed between gold electrodes and the electron transport is measured (phthalocyanine ligands are good conductors)
- Experiment investigates how electronic and nuclear spin states influence the transport
- Variable magnetic field applied



Easy axis

[1] Vincent et. al., Nature, 2012, 488, 357; [2] Thiele et. al., Science, 2014, 344, 1135

- The molecule is kept at ~ 0.08 K and so only ground electronic states are relevant, which are  $m_J = \pm 6$
- $m_J$  $m_I$ • Tb has 100% isotopic abundance of <sup>159</sup>Tb with I = 3/2QTM +6,+3/2> When molecule is +6, +1/2+6,-1/2> magnetised, one of +6,-3/2> four states are occupied E (K) 0 Avoided crossings near 6 - 3/2B = 0 allows magnetic relaxation -80 80 0 B (mT)

[1] Vincent et. al., Nature, 2012, 488, 357; [2] Thiele et. al., Science, 2014, 344, 1135

• The electron transport (i.e. conductivity) of the molecule is measured



[1] Vincent et. al., Nature, 2012, 488, 357; [2] Thiele et. al., Science, 2014, 344, 1135

- The conductivity jumps when the magnetisation relaxes
- This directly measures the relaxation of a single molecule (multiple scans needed to see the different steps)
- Nuclear spin states live long enough to control coherently...





• Nuclear spin is initialised in the  $m_I = +3/2$  state with magnetic field

• Microwave pulse induces oscillations between  $m_I = +3/2$  and  $m_I = +1/2$ 

• Sweep magnetic field back to read-out *m<sub>I</sub>* state



Vincent et. al., Nature, 2012, 488, 357; Thiele et. al., Science, 2014, 344, 1135

#### Grover's algorithm in a molecule



[1] C. Godfrin et al., Phys. Rev. Lett., 2017, 119, 187702.

## Building a quantum computer

- The DiVincenzo criteria:
- 1. Scalable system with well-characterized qubits
- 2. The ability to initialize the qubits
- 3. Long coherence times
- 4. A universal set of quantum gates
- 5. Qubit readout ability
- Many ways to implement qubits:
  - Trapped ions
  - Superconducting circuits (D-Wave, Google, IBM)
  - Defects in diamond
  - Molecular magnets

## Qubits



[1] E. J. L. McInnes *et al.*, *Angew. Chem. Int. Ed.*, 2015, 54, 14244; [2] C. J. Wedge et al., *Phys. Rev. B*, 2012, 108, 107204;
[3] J. Ferrando-Soria *et al.*, *Nature Commun.*, 2016, 7, 11377.